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Units and Measurements (Dimensional Analysis & Significant Figures) Contact Number: 9667591930 / 8527521718

1.

A physical quantity X is expressed as

$$_{V}=\sqrt{\frac{X}{\mathrm{Density}}}$$

The dimensions of X are

- 1. $ML^{-1}T^{-2}$
- 2. MLT⁻²
- 3. MLT^{-1}
- 4. $ML^{-1}T^{-3}$

2.

For $10^{(at+3)}$, the dimension of a is :

- $1.\ \mathrm{M^0L^0T^0}$
- 2. $M^0L^0T^1$
- 3. $M^0L^0T^{-1}$
- 4. None of these

3.

If force (F), velocity (v) and time (T) are taken as fundamental units, then the dimensions of mass are:

- (a) [FvT⁻¹]
- (b) [FvT⁻²]
- (c) $[Fv^{-1}T^{-1}]$
- (d) $[Fv^{-1}T]$

4.

The Martians use force (F), acceleration (A) and time (T) as their fundamental physical quantities. The dimensions of length on Martians system are

- $(1) [FT^2]$
- (2) $[F^{-1}T^2]$
- (3) $[F^{-1}A^2T^{-1}]$
- $(4) [AT^2]$

5.

Which of the following quantities has the same dimensions as that of energy ?

- (1) Power
- (2) Force
- (3) Momentum
- (4) Work

6.

Position of a body with acceleration 'a' is given by $x = Ka^m t^n$, here t is time. Find values of m and n.

- 1. m = 1, n = 1
- 2. m = 1, n = 2
- 3. m = 2, n = 1
- 4. m = 2, n = 2

7.

Unit of power is:

- (1) Kilowatt
- (2) Kilowatt-hour
- (3) Dyne
- (4) Joule

8.

Taking into account the significant figures, what is the value of (9.99m - 0.0099m)?

- 1. 9.98 m
- 2. 9.980 m
- 3. 9.9 m
- 4. 9.9801 m

9.

Which of the following measurements is most precise?

- 1. 5.00 mm
- 2. 5.00 cm
- 3. 5.00 m
- 4. 5.00 km

10.

What is the number of significant figures in 0.310×10^3 ?

- (1) 2
- (2)3
- (3)4
- (4)6

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11.

One nanometre is equal to:

- $(1)\ 10^9 mm$
- (2) 10^{-6} cm
- (3) $10^{-7}cm$
- $(4) 10^{-9} cm$

12.

If 97.52 is divided by 2.54, the correct result in terms of significant figures is:

- (1)38.4
- (2) 38.3937
- (3) 38.394
- (4) 38.39

13.

The measurement are made as 18.425 cm, 7.21 cm and 5.0 cm. The addition

should be written as:

- (a) 30.635 cm
- $(b)\ 30.\ 64\ cm$
- (c) 30.63 cm
- $(d) \ 30.6 \ cm$

14.

Choose any one of the following four responses:

- (1) If both assertion and reason are true and the reason is the correct explanation of the assertion.
- (2) If both assertion and reason are true but reason is not the correct explanation of the assertion.
- (3) If assertion is true but reason is false.
- (4) If the assertion and reason both are false.

Assertion: Dimensional constants are the quantities whose values are constant.

Reason: Dimensional constants are dimensionless.

15.

The decimal equivalent of 1/20 up to three significant figures is:

- (1) 0.0500
- (2) 0.05000
- (3) 0.0050
- $(4) 5.0 \times 10^{-2}$

16.

The number of significant figures in all the given numbers 25.12, 2009, 4.156 and 1 .217 $\times 10^{-4}$ is:

- 1. 1
- 2.2
- 3.3
- 4.4

17.

The sum of the numbers 436.32, 227.2, and 0.301 in appropriate significant figures is:

- (a) 663.821
- (b) 664
- (c) 663.8
- (d) 663.82

18.

The length, breadth, and thickness of a block are given by l = 12 cm, b = 6 cm and t = 2.45 cm The volume of the block according to the idea of significant figures should be:

- $(1) 1 \times 10^2 \text{ cm}^3$
- (2) $2 \times 10^2 \text{ cm}^3$
- (3) $1.764 \times 10^2 \text{ cm}^3$
- (4) None of these

19.

The numbers 2.745 and 2.735 on rounding off to 3 significant figures will give:

- (a) 2.75 and 2.74
- (b) 2.74 and 2.73
- (c) 2.75 and 2.73
- (d) 2.74 and 2.74



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20.

The mass and volume of a body are 4.237 g and 2.5 cm^3 , respectively. The density of the material of the body in correct significant figures is:

- (a) 1.6048 g cm^{-3}
- (b) 1.69 g cm^{-3}
- (c) 1.7 g cm^{-3}
- (d) 1.695 g cm^{-3}

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